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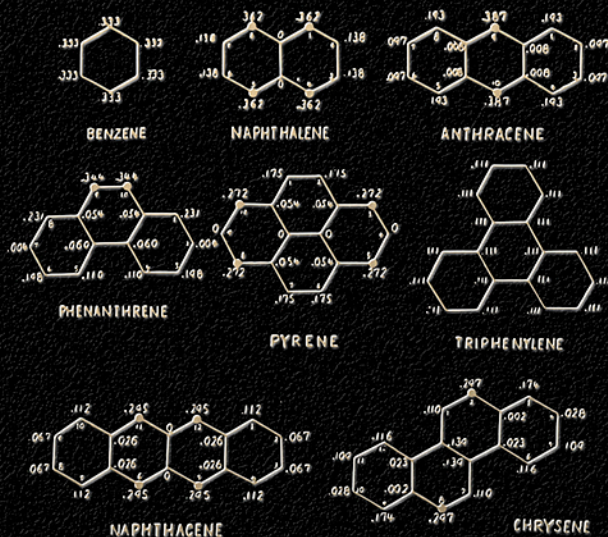
A Molecular Orbital Theory of Reactivity in Aromatic Hydrocarbons

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In the present paper it is shown that on the ground of the latter method, if we distinguish the pair of π -electrons in the highest occupied orbital in the ground state from the others and assume that this pair of π -electrons plays a decisive role in the reaction in question, we can obtain an illuminating explanation of the difference of reactivity at each position in a molecule.

In the first place we will report on the treatment of unsubstituted aromatic hydrocarbons. The orienting effect of substituents in aromatic nuclei will be treated in the next publication.



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